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$N_1 + N_2 + \dots$  It follows that the sum of all the mole fractions of the mixture must also equal 1.  $X_1 + X_2 + \dots = 1$  The mass of component  $i$  in a mixture is the product of the number of moles of  $i$  and its molecular weight,  $M_i$ . The mass of the mixture is therefore the sum,  $M = N_1M_1 + N_2M_2 + \dots$

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